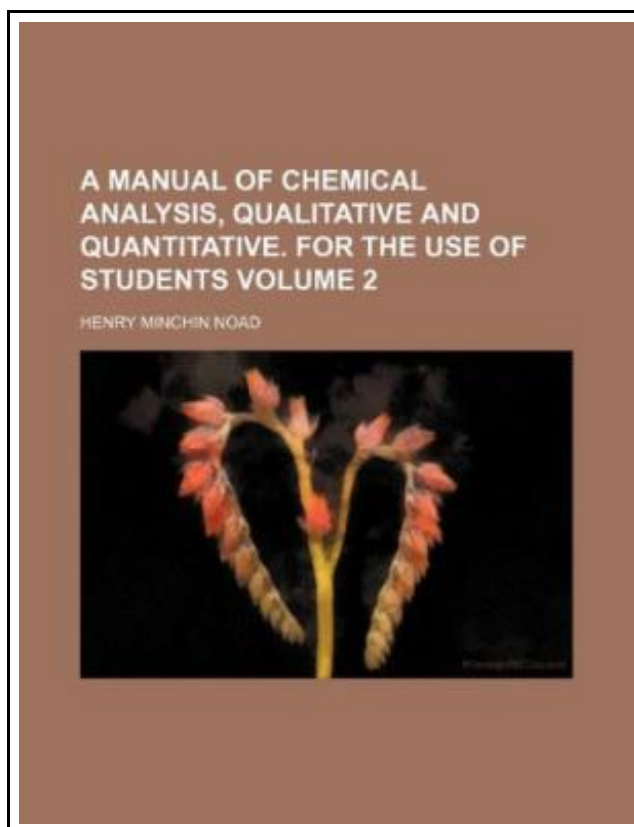


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1864 Excerpt: .into four equal portions. I. The sulphuric acid is determined by chloride of barium, the precipitated sulphate of baryta being washed first for some time with boiling distilled water, and then with water rendered slightly acid with hydrochloric acid; the solution must not be acid previous to adding the chloride of barium, or a partial oxidation of the sulphurous acid may take place, thus occasioning too large an amount of sulphate of baryta. li. Another portion is mixed with a considerable quantity Six of these compounds have been well made out, and the existence of several others is probable; the six well-defined acids have the following composition: --Sojjsojssosjos.,)!, and they have been thus classified by Berzelius: -- Monothionic S03, S02. Sulphuric and sulphurous acids. Dithionic. Sjo.SoOs Hyposulphurous and hyposulphuric acids. Trithionic. S305. Sulphyposulphuric acid of Langlois. Tetrathionic. S405. Bisulpho-hyposulphuric acid of Fordos and Gelis. Besides these, a new series of acids of sulphur has been announced by Plessy (Comptes Rendus, Aug. 1845), to which he assigns the following composition: --S506, S50r, SfiO;, S8010; and M. Wackenroder, and MM. Fordos and Gelis, have each described an acid having the composition S50s. of carbonate of magnesia, and saturated with iodine, the quantity required being carefully noted; the amount of sulphuric acid contained in the liquid is again determined by chloride of barium; the quantity of sulphate of baryta obtained will be greater than in the first experiment, and the increase will indicate the quantity of sulphurous acid,...

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